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10 SACE 2 Essentials Chemistry WorkbookCHAPTER 1Group VI elementsIn their compounds –either they exhibit a covalence of 2, as in the case of oxygen in all of its compounds such as H 2 O, and of sulfurand selenium in some of their compounds, for example SF 2 . In these compounds the oxygen, sulfur andselenium atoms share only two of the p 4 electrons from the s 2 p 4 outer shell configuration to formcovalent bonds with other non-metal atoms.

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